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A useful Preparation of Pyrroles from α,β-unsaturated Sulfones

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Abstract: The addition of alkyl isocyanoacetate to α,β -unsaturated sulfones, which are easily obtained from olefins, affords a convenient access to pyrroles with unusual substitution patterns.

Pyrrole itself and substituted pyrroles are important starting materials for the synthesis of naturally occuring porphinoid compounds¹ and other porphyrins. For instance, the synthesis of new types of porphyrin derivatives with altered or extended chromophores² or unusual substitution patterns³ affords new pyrrolic subunits.

In the course of our studies directed towards the synthesis of new chlorins (dihydroporphyrins) the pyrroles 4 and 8 were required as building blocks. Obviously none of the various classical pyrrole syntheses⁴ were suitable for our purposes to prepare *rac*-4 and 8. The methods of van Leusen⁵ and Barton⁶, which are based on the work of Schöllkopf⁷ and have become important tools in pyrrole synthesis, led to the solution of our synthetic problem.

(a)
$$H + C'$$
 $Tos \sim N^{c'}$
 $H \rightarrow C'$
 $H \rightarrow C'$

Scheme 1: Synthesis of pyrroles according to (a) van Leusen⁵ and (b) Barton⁶.

According to van Leusens method pyrroles are obtained by addition of tosylmethyl isocyanide (TosMIC) to α,β -unsaturated carbonyl compounds in the presence of base. Mechanistically the pyrrole ring is formed by 1.4-addition of the deprotonated isocyanide to the unsaturated carbonyl compound followed by nucleophilic attack of the enolate anion to the isocyanide carbon atom. Elimination of the tosyl group and tautomerisation yields the aromatic system (s. scheme 1a). Barton's method makes use of alkyl isocyanoacetates and nitroolefins instead of TosMIC and α,β -unsaturated carbonyl compounds. In this case the nitro function makes 1.4-addition possible, stabilizes the intermediate anion and serves as leaving group to yield the pyrrole (s. scheme 1b).

EtO₂C
$$\rightarrow$$
 EtO₂C \rightarrow SO₂Ph \rightarrow EtO₂C \rightarrow SO₂Ph \rightarrow EtO₂C \rightarrow NH EtO₂C \rightarrow SO₂Ph \rightarrow EtO₂C \rightarrow SO₂Ph \rightarrow EtO₂C \rightarrow SO₂Ph \rightarrow CO₂Et \rightarrow SO₂Ph \rightarrow O \rightarrow SO₂Ph \rightarrow O \rightarrow SO₂Ph \rightarrow O \rightarrow SO₂Ph \rightarrow O \rightarrow NH CO₂Et \rightarrow SO₂Ph \rightarrow O \rightarrow SO₂Ph \rightarrow O \rightarrow SO₂Ph \rightarrow

Scheme 2: a: 1) 1.1 eq. PhSCl, CH₂Cl₂, -15°C to rt, 20 min; 2) 1.9-3 eq. Oxone[®], H₂O, MeOH, rt 1-2 h. b: 1.1 eq DBU, CH₂Cl₂, 0°C to rt, 0.5-1 h. c: 1) 1.1 eq CNCH₂CO₂Et, THF, rt; 2) 2.2 eq. t-BuOK, THF, rt, 3-12 h.

The desired pyrroles rac-4 and 8 are obtained from the α,β -unsaturated sulfones rac-3 and 7 by addition of alkyl isocyanoacetates in the presence of base⁸. The phenylsulfonyl group has the same function as the nitro group in Barton's method, but has the advantage that it may be introduced more conveniently. For this purpose the readily available olefins $rac-1^9$ and 5^{10} were treated with phenylsulfenyl chloride, followed by oxidation of the sulfides to the corresponding sulfones rac-2 and rac-6. The phenylsulfonyl function allows elimination under mild conditions to yield the α,β -unsaturated sulfones rac-3 and 7. The described method should provide a general access to pyrroles starting from readily available olefins.

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